ame:		Date:	F	Period:
1. Complete the	e names of the	following ic	ons.	
3a <sup>2+</sup>	Al <sup>3+</sup>	K⁺	N <sup>3-</sup>	O <sup>2-</sup>
-	P <sup>3-</sup>	S <sup>2-</sup>	CI <sup>-</sup>	—— Na <sup>+</sup>
-			and a	-
2. Write the nar	mes of the follo	wing comp	ounds.	
a. CaO			_	
b. KBr			_	
c. Al <sub>2</sub> O <sub>3</sub>	-			
d. MgCl2			_	
3. Write the for	mulas and nam	es for com	ounds of the fall	louing ions.
o. Write the for	maias ana nan	ies for comp	odinas of the for	lowing ions.
	Br <sup>-</sup>		S <sup>2-</sup>	N <sup>3-</sup>
Na⁺				
			1	
Al <sup>3+</sup>				

Name:	Date:		Period:
4. Select the co	orrect name for each.		
A. Fe <sub>2</sub> S <sub>3</sub>			
1) ir	on sulfide		
2) ir	on(II) sulfide		
	on(III) sulfide		
B. CuO			
1) co	pper oxide		
2) co	pper(I) oxide		
3) co	pper(II) oxide		
5. What is the	correct formula for each	n of the following	?
A. Copper(I)	nitride		
1) Cu	N 2) CuN₃	3) Cu <sub>3</sub> N	
B. Lead(IV)	oxide		
1) Pb		3) Pb <sub>2</sub> O <sub>4</sub>	